SUPPORT-INDUCED EFFECTS ON THE IRIDIUM NANOPARTICLES ACTIVITY, SELECTIVITY AND STABILITY PERFORMANCE UNDER THE CO₂ REFORMING **OF METHANE REACTION**

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The dry (CO₂) reforming of methane (DRM, Eq.1) for syngas production – a critical feedstock for the production of hydrogen, ammonia and liquid energy carriers- ranks among the top issues of applied catalysis in the light of environmental protection, renewable energy production and circular economy [1,2]. DRM involves the simultaneous reduction of two key greenhouse gases (CO2 and CH4) and provides an efficient way for CO₂ recycling as well as the direct utilization of biogas. Among others, these advantages make DRM (Eq. 1) a more favorable process compared to the other alternatives, steam reforming (SRM, Eq. 2) and oxy reforming (POM, Eq. 3):

$DRM: CH_4 + CO_2 \leftrightarrow 2CO + 2H_2$		$\Delta H^{\circ} = 247 \text{ kJ} \cdot \text{mol}^{-1}$	(Eq. 1)
SRM:	$CH_4 + H_2O \leftrightarrow CO + 3H_2$	$\Delta H^{\circ} = 206 \text{ kJ} \cdot \text{mol}^{-1}$	(Eq. 2)
POM:	$CH_4 + \frac{1}{2} O_2 \leftrightarrow CO + 2H_2$	$\Delta H^{\circ} = -36 \text{ kJ} \cdot \text{mol}^{-1}$	(Eq. 3)

Here we report on the effect of the metal oxide supports (Y-Al₂O₃, alumina-ceria-zirconia (ACZ) and ceria-zirconia (CZ)) on the low temperature (ca. 500-750 °C) DRM activity, selectivity, resistance to carbon deposition as well as on the stability under high temperature oxidative aging of Ir nanoparticles dispersed on them. A variety of characterization techniques were implemented to provide significant insight into the factors (e.g. metal-support interactions and materials properties) that determine Ir intrinsic kinetics and stability during DRM. It was found that all Ir/γ-Al₂O₃, Ir/ACZ and Ir/CZ catalysts have a very stable time-on-stream DRM performance, although supports with high oxygen storage capacity (i.e. ACZ and CZ) promoted CO₂ conversion, yielding CO-enriched syngas. For all catalysts carbon deposition was low, although it is decreasing in the order Ir/y-Al₂O₃>Ir/ACZ>Ir/CZ that is consistent with a bifunctional mechanism involving participation of oxygen vacancies on the surface of the support in CO₂ activation and carbon removal. The lower apparent activation energy for CO₂ consumption rate observed with CZ-containing catalysts (Ir/ACZ and Ir/CZ) suggests that CZ is a promising support for use in low temperature DRM.

Keywords: Dry reforming of methane; Ceria-zirconia mixed oxides; Iridium nanoparticles; resistance to carbon deposition.

References

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