

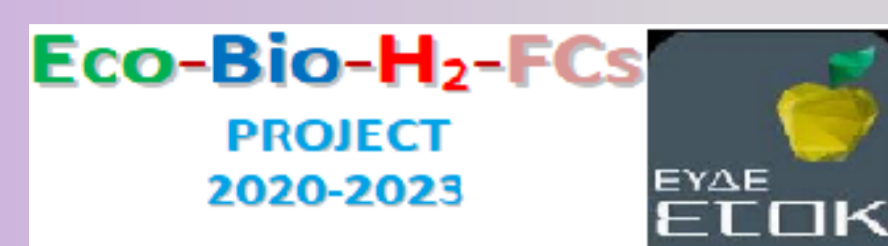
Catalytic performance and stability of Ru on Ce-based Aminoclay carriers for Sabatier reaction

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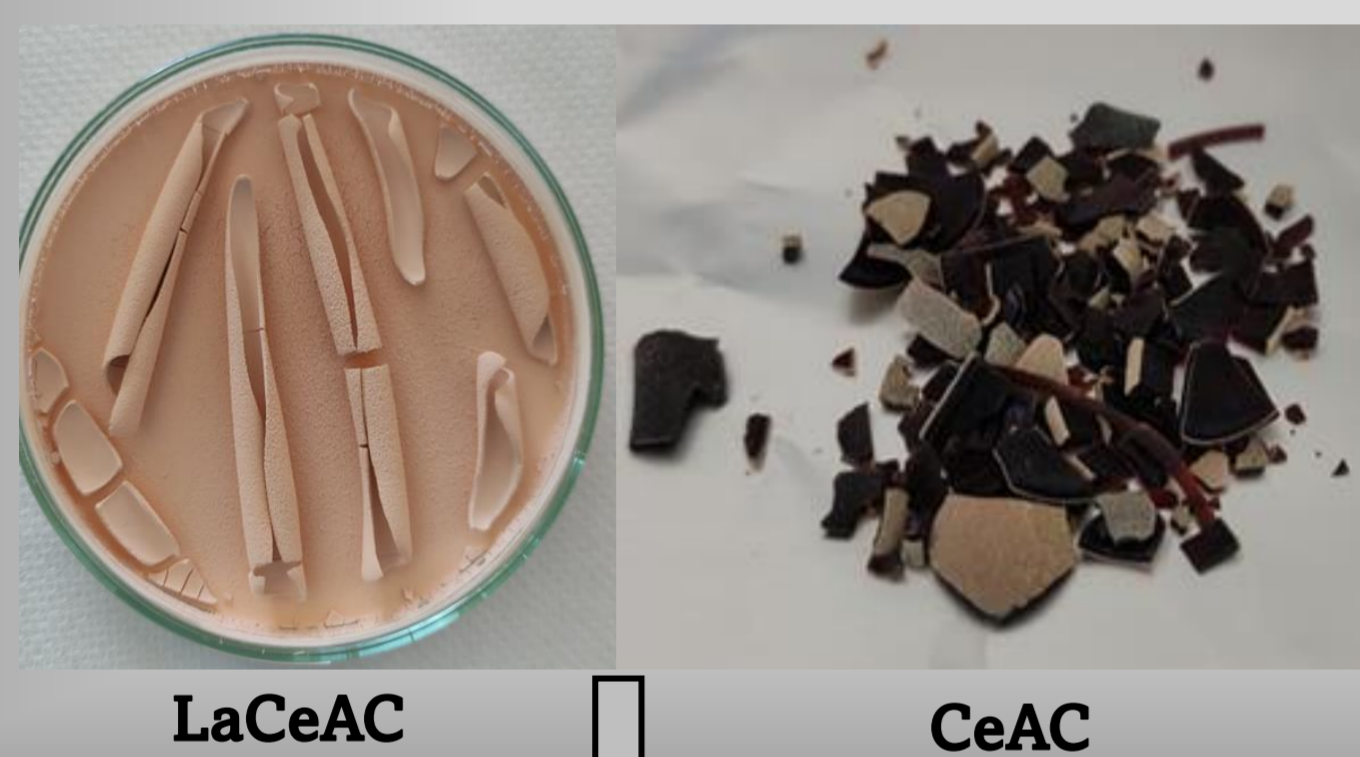


Motivation

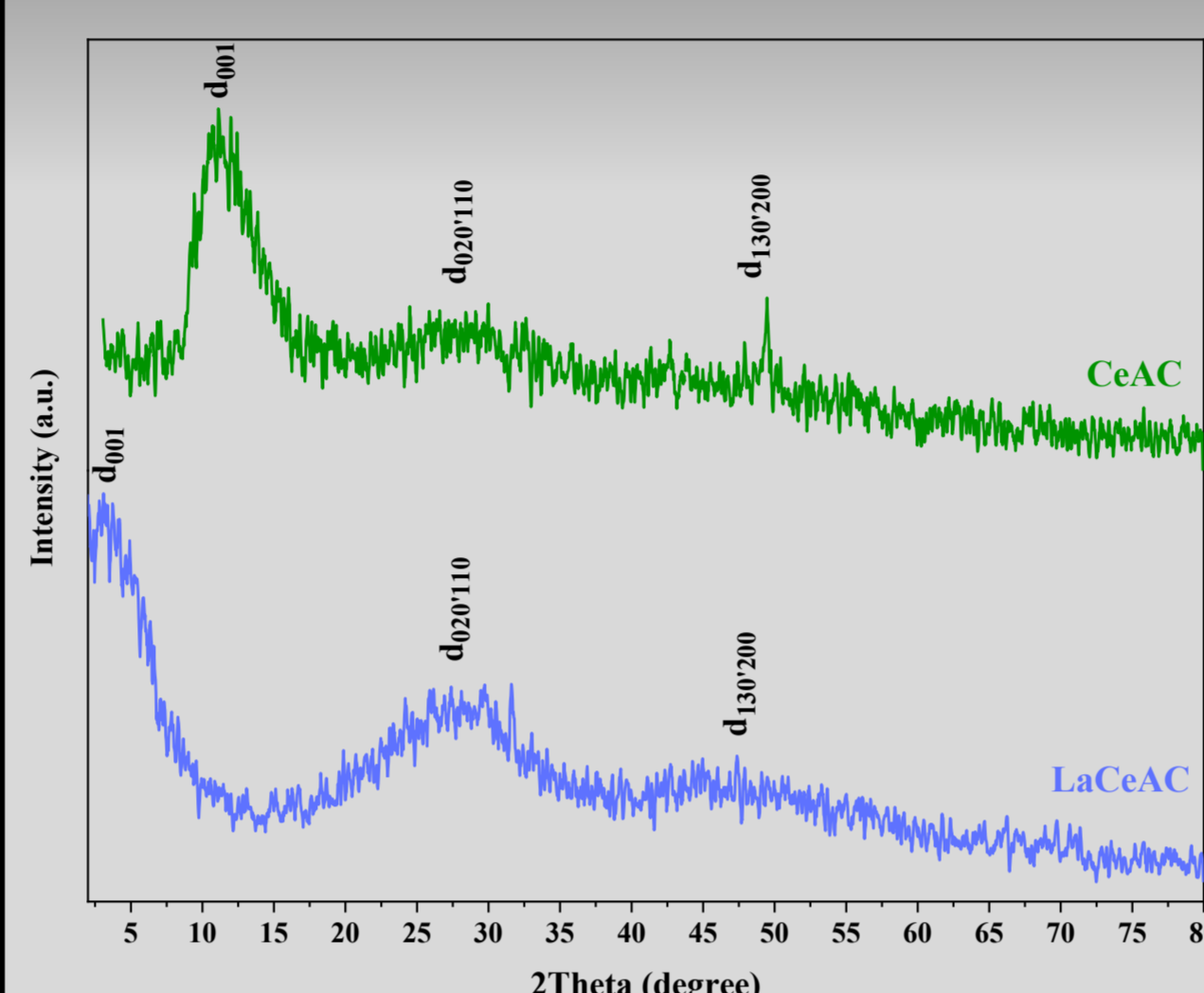
Carbon dioxide (CO₂) is a colorless, and odorless greenhouse gas (GHG) with a faint acid taste. Over the past years, CO₂ concentrations in atmosphere have increased by 30%. As a result, the global surface temperature increased from 0.4 to 0.8 °C. The catalytic hydrogenation of CO₂ to produce CH₄, also known as Sabatier reaction (CO₂ + 4H₂ ↔ CH₄ + 2H₂O; ΔH₀ = -164.7 kJ/mol) is considered an extremely important route for CO₂ recycling, with Ni and Ru are among the most active Sabatier catalysts. In this regard, cerium and lanthanum/cerium based synthetic Aminoclay analogues were prepared, using a green, facile, and cost-effective room-temperature sol-gel-based synthetic method, as carriers of Ru nanoparticles (Ru/CeAC, Ru/LaCeAC). The performance of this new class of catalysts was investigated under CO₂ hydrogenation conditions were found to be highly active and selective towards CH₄ production.

Design of CeAC & LaCeAC carriers

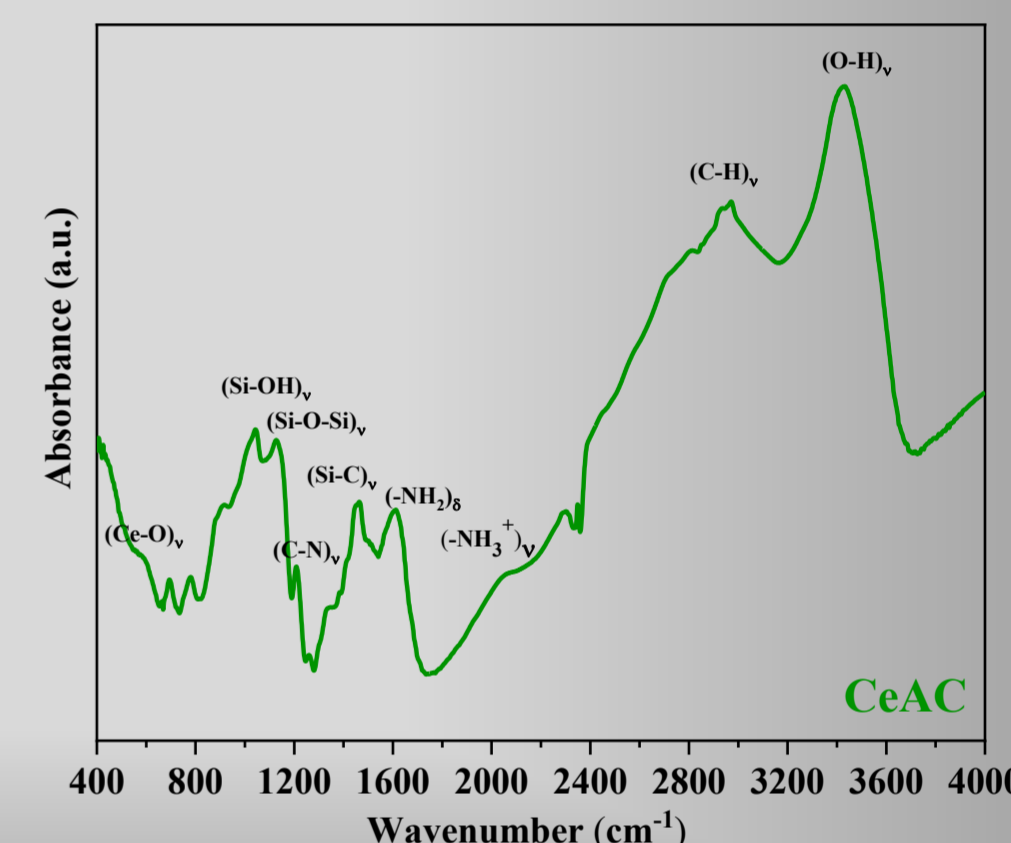
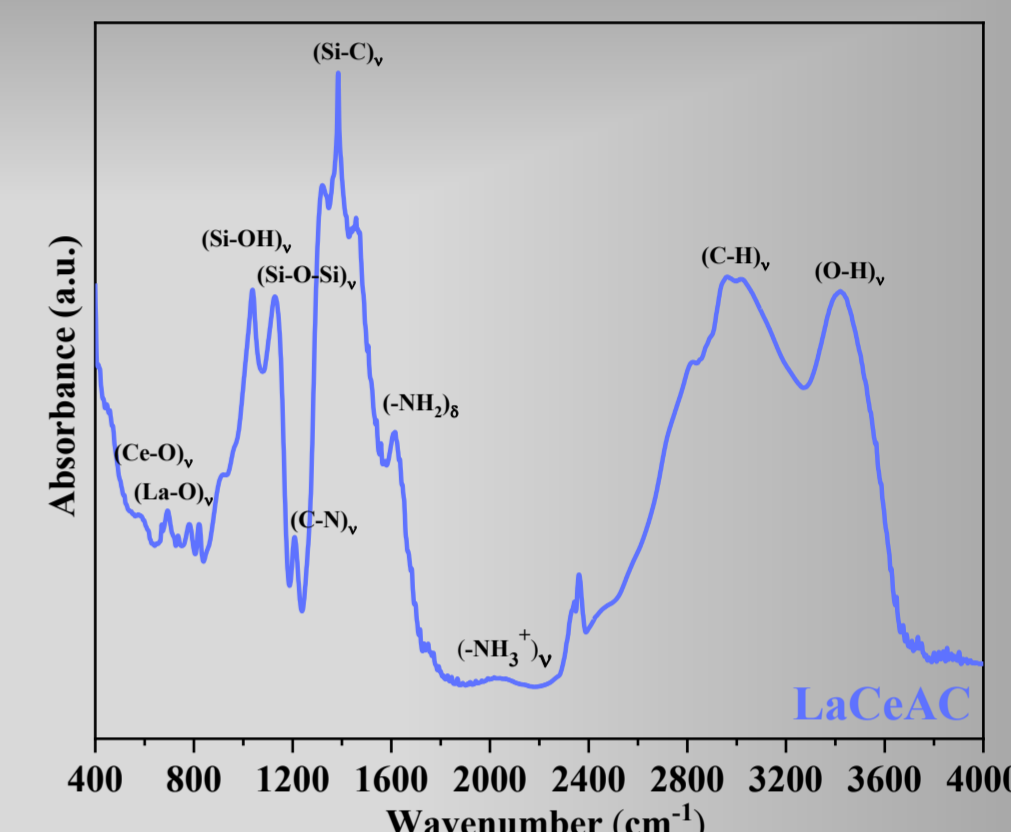
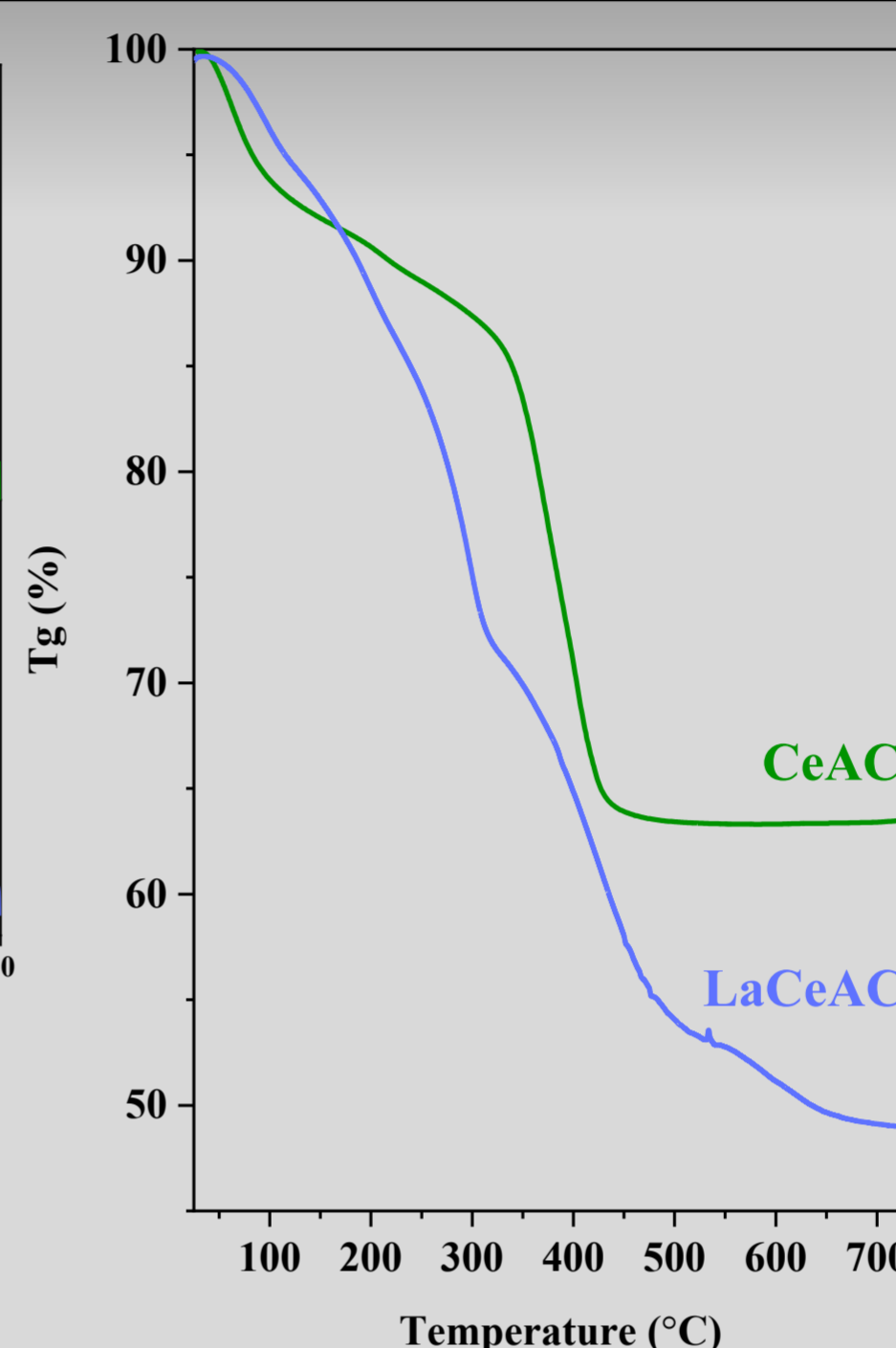
For the preparation of lanthanum-cerium synthetic aminoclay (LaCeAC), 1.81 mmol of CeCl₃ · 7H₂O and 1.81 mmol of La(NO₃)₃ · 6H₂O were added in 20 ml ethanol, followed by stirring at room temperature. After 10 min of stirring, 5.85 mmol of N-[3-(Trimethoxy silyl) propyl] ethylenediamine, was added dropwise. The solution remained under continuous stirring for 24 h. A gel-type material obtained after reaction was recovered by centrifugation followed by washing with ethanol. Finally, the sample dried in the oven at 40 °C overnight. Cerium synthetic aminoclay (CeAC) was also prepared according to the previous procedure using 3.62 mmol Cerium chloride heptahydrate (CeCl₃ · 7H₂O).



Structural characterization



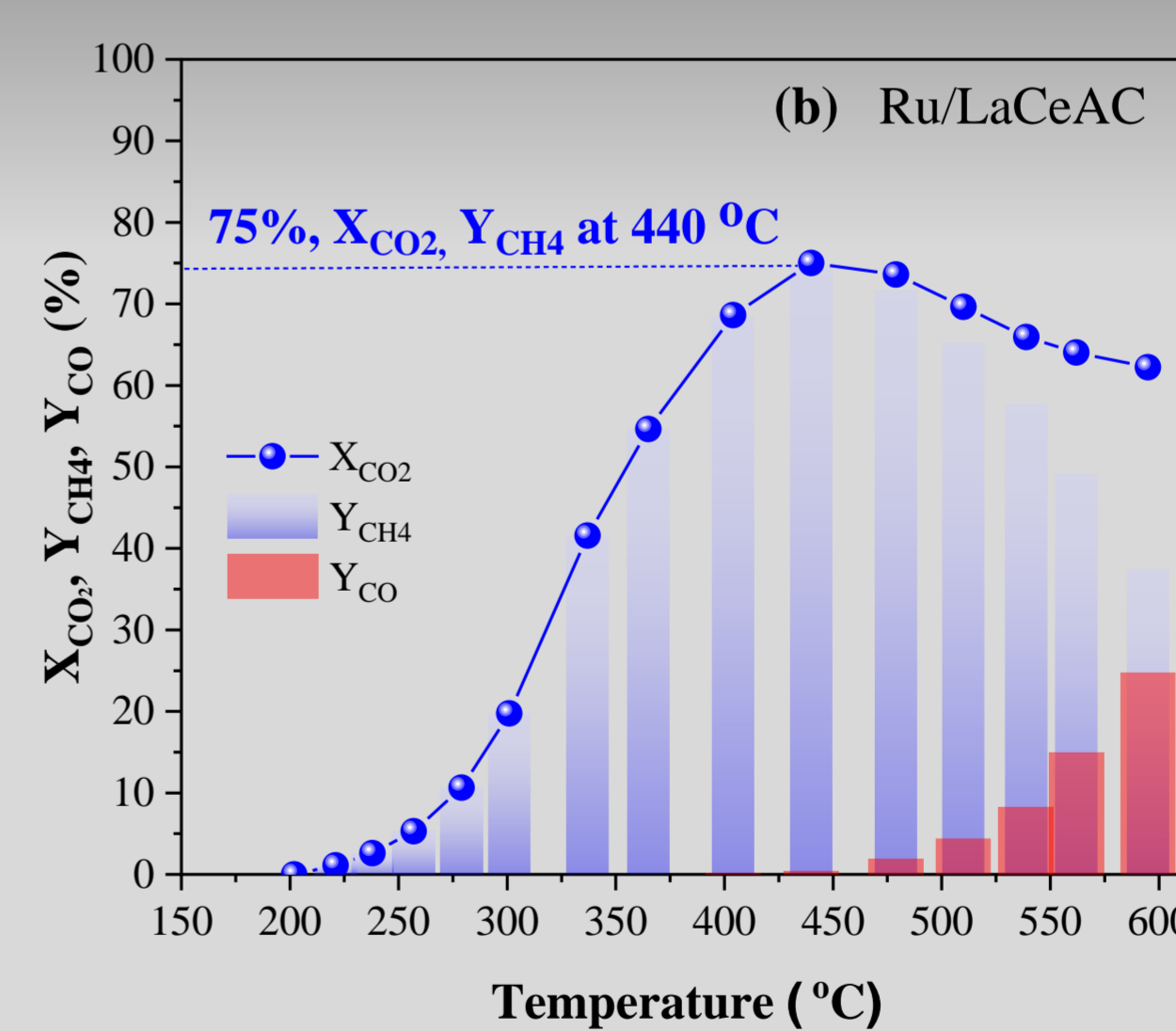
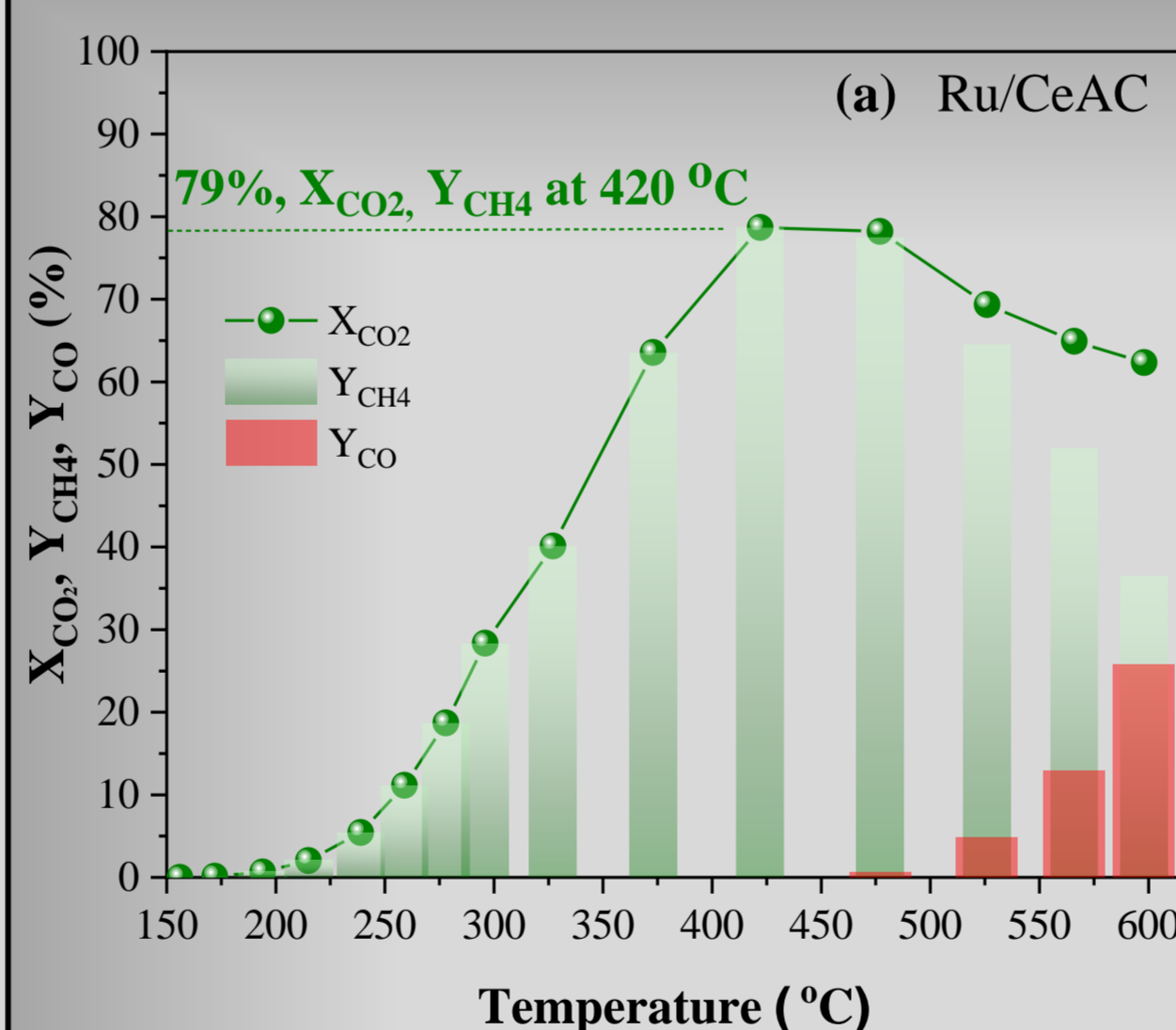
The basal spacing at d₀₀₁ = 27 Å for the LaCeAC and d₀₀₁ = 8 Å for CeAC, respectively, indicates the thickness of the regularly layered unit clay structure with a bilayer arrangement of propylamino groups.



Design of Ru/CeAC and Ru/LaCeAC catalysts

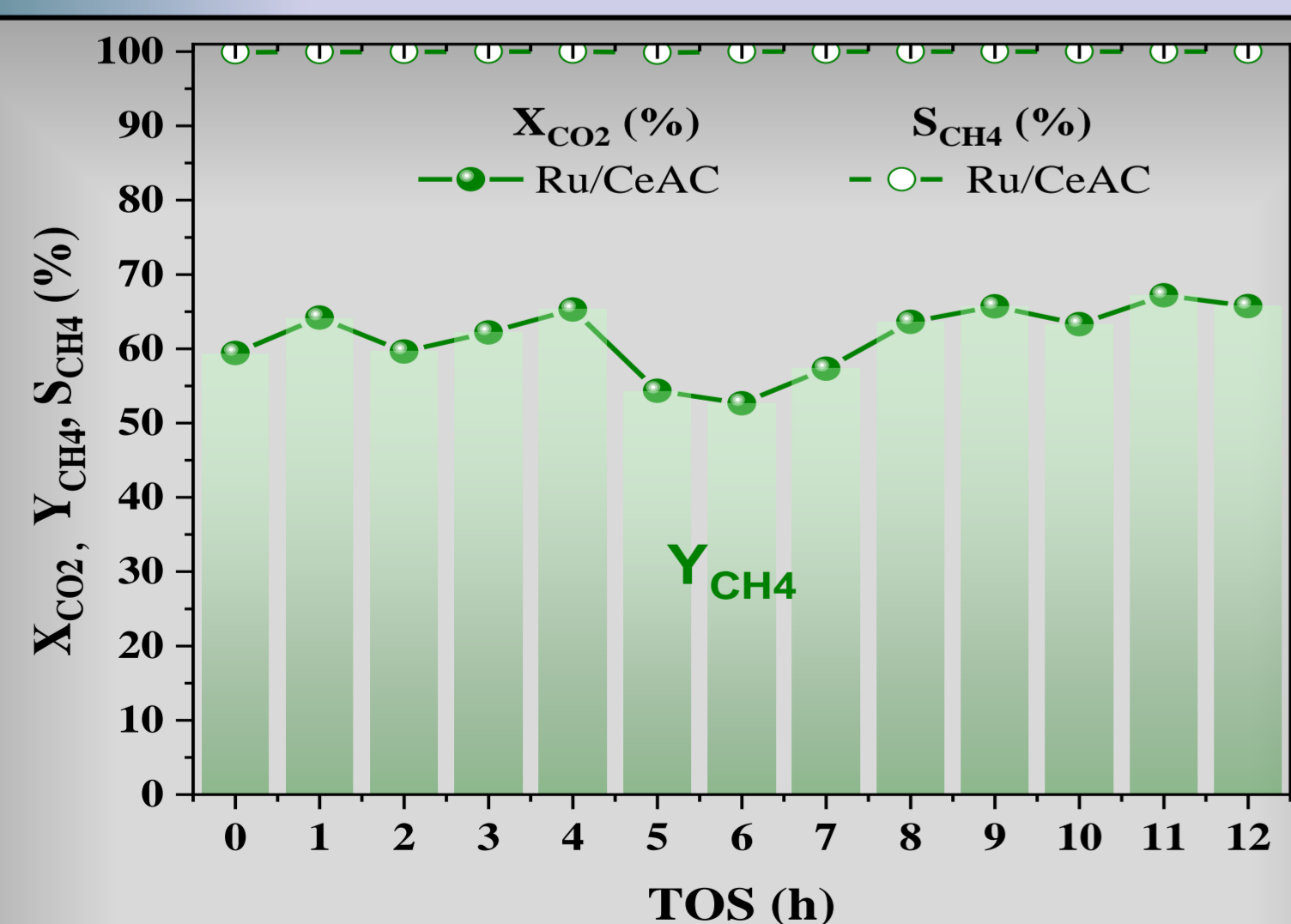
Ruthenium was incorporated on Ce-based aminoclay supports using the conventional wet impregnation method. Appropriate amounts of Ce-based aminoclay were impregnated by specific volume of a Ruthenium (III) nitrosyl solution to obtain a Ru loading of 3%. Then, pH was adjusted to a value of 6 by adding NH₃ solution. The slurry was dried under continuous stirring at 80 °C. The resulting material was further dried at 110 °C overnight and then was calcinated at 450 °C for 1 h.

Ru/CeAC and Ru/LaCeAC catalysts



CO₂ conversion (X_{CO₂}), CH₄ and CO yield (Y_{CH₄} & Y_{CO}) for catalysts Ru/CeAC (a) and Ru/LaCeAC (b) versus temperature. Experimental Conditions: 25% H₂, 5% CO₂ in balance with Ar, 1 atm, T (°C): 150-600 °C, F_t = 19 cc/min, m_{cat} = 60 mg.

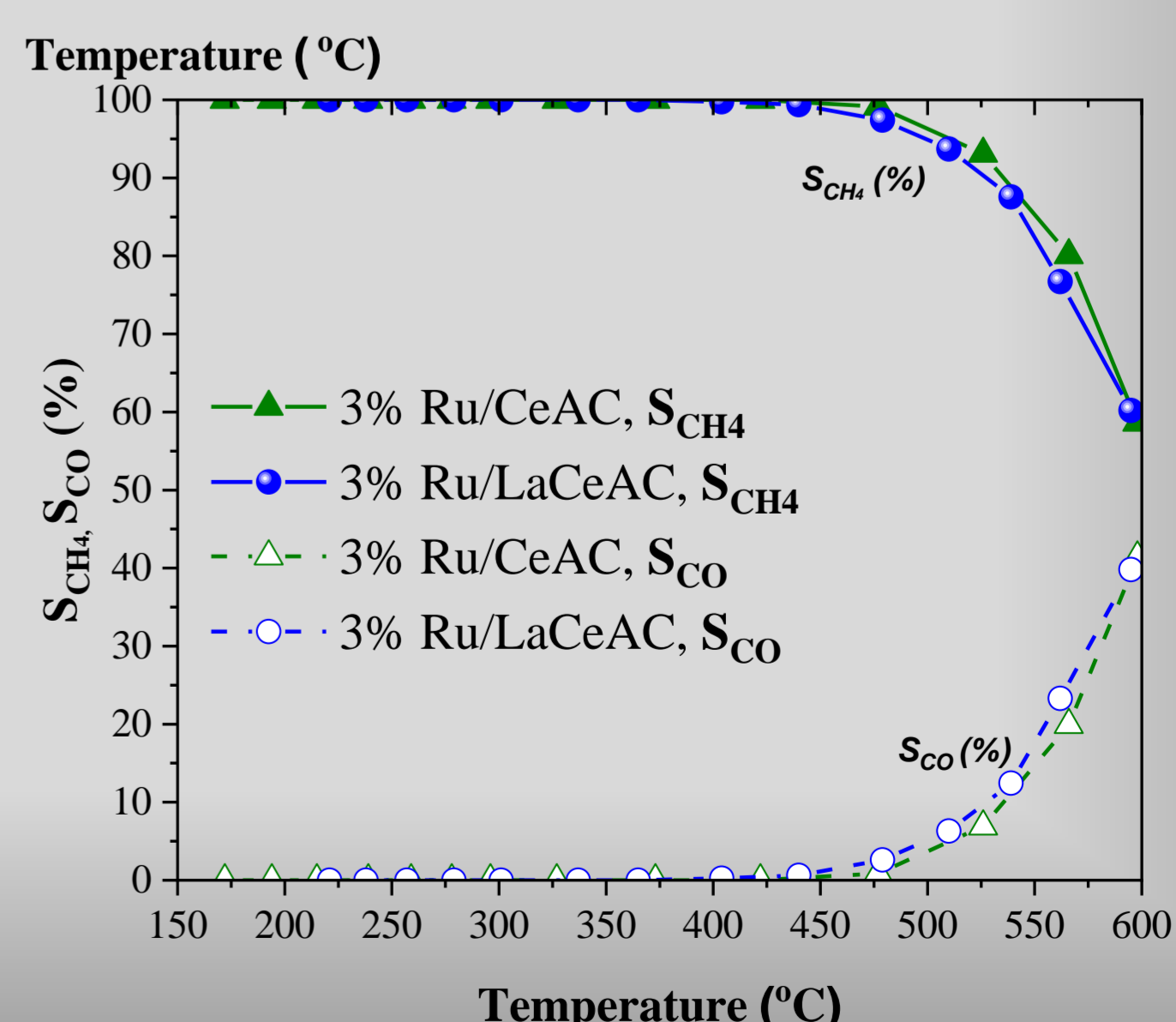
Conclusions



CO₂ conversion (X_{CO₂}), CH₄ yield and selectivity (Y_{CH₄} & S_{CH₄}) for catalysts Ru/CeAC versus TOS (Time-on-stream, h). Experimental Conditions: 25% H₂, 5% CO₂ in balance with Ar, 1 atm, T (°C): 380 °C, F_t = 19 cc/min, m_{cat} = 60 mg.

The up to 70% CH₄ yield obtained reveals that synthetic Ce-based Aminoclays can be successfully used as supports for the design of efficient CO₂ methanation catalysts.

CH₄ and CO selectivity (S_{CH₄} & S_{CO}) for catalysts Ru/CeAC and Ru/LaCeAC versus temperature. Experimental Conditions: 25% H₂, 5% CO₂ in balance with Ar, 1 atm, T (°C): 150-600 °C, F_t = 19 cc/min, m_{cat} = 60 mg.



References

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Acknowledgements

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